Research Article



Investigation of carbon nano-tube (CNT) particles effect on the performance of a refrigeration cycle

Bahram Jalili^{a,*}, Hamed Ghafoori^b, Payam Jalili^c

^a Young Researchers and Elite Club, Ayatollah Amoli Branch, Islamic Azad University, Amol, Iran

^b Department of Mechanical Engineering, Takestan Branch, Islamic Azad University, Takestan, Iran

^c Young Researchers and Elite Club ,Ayatollah Amoli Branch, Islamic Azad University, Amol, Iran

* Corresponding author. Tel.: +989124031827; E-mail address: bahramjalily@yahoo.com

Abstract

	In spite of previous research in Air Condition, now days because of improvement
Keywords:	in time and cost we should apply modern technology such as nano technology in
Refrigeration cycle,	this part of science and research. In this paper, the performance and Heat transfer
Carbon nanotube	of a refrigeration cycle with nanoparticles in the working fluid was investigated
particles,	experimentally. Water with nanoparticles mixtures were used in this experiment.
Heat transfer,	The tests with carbon nano tube nanoparticles showed that the different particle
Performance.	volume fractions have different effect on the refrigerator performance. According to results, it is shown that increasing in CNT particles will be caused great temperature gradient in entry and discharging evaporator respect to pure water. Thus, nanoparticles can be used in refrigeration cycle to considerably increase heat transfer.

Accepted:23 February2014 © Academic Research Online Publisher. All rights reserved.

1. Introduction

Improvement of thermal conductivity of liquids was earlier made possible by mixing micronsized particles with a base fluid [1]. But, because of some problem such as rapid sedimentation, erosion, clogging and high-pressure drop caused by these particles has kept the technology far from practical use. When nano particles dispersed uniformly and suspended stably in base fluids, can provide impressive improvements in the thermal properties of base fluids. Nanofluids, which are a colloidal mixture of nanoparticles (1– 100 nm) and a base liquid (nanoparticle fluid suspensions) are the term first coined by Choi in the year 1995 [2] at the Argonne National Laboratory. Nanoparticles Compared to micron-sized particles, are engineered to have larger relative surface areas, less particle momentum, high mobility, better suspension stability than micron-sized particles and importantly increase the thermal conductivity of the mixture. Nano fluids are favorable working fluids in medium

as coolants, lubricants, hydraulic and metal cutting. Further, a negligible pressure drop and mechanical abrasion makes researchers subscribe to nanofluids for the development of the next generation miniaturized heat exchangers. Based on where we want to use, nanoparticles have been made of various materials [3–17] such as oxide ceramics, nitride ceramics, carbide ceramics, metals, semiconductors, carbon nanotubes and composite materials.

2. Basic Refrigeration Cycle

Refrigeration is a process in which work is done to move heat from one location to another. The work of heat transport is traditionally driven by mechanical work, but can also be driven by heat, magnetism, electricity, laser, or other means. Refrigeration has many applications, including, but not limited to: household refrigerators, industrial freezers, cryogenics, and air conditioning. Heat pumps may use the heat output of the refrigeration process, and also may be designed to be reversible, but are otherwise similar to refrigeration units. Although the entire chiller package is more complex, the basic components required for mechanical refrigeration are the compressor, evaporator, condenser and thermostatic expansion valve.



Fig. 1: T-s diagram of refrigeration cycle.



Fig. 2: P-h diagram of refrigeration cycle.

Table 1: Refrigeration process

Process	Description					
1-2s	A reversible, adiabatic (isentropic) compression of the refrigerant. The saturate vapor at state 1 is superheated to state 2.					
	$\Rightarrow W_c = h_{2s} - h_1$	(1)				
2s-3	An internally, reversible, constant substance is desuperheated and ther process, the working substance reje- water.	pressure heat rejection in which the working a condensed to a saturated liquid at 3. During his texts most of its energy to the condenser cooling				
	$\Rightarrow q_h = h_{2s} - h_2$	(2)				
3-4	An irreversible throttling process in which the temperature and pressure decrease at constant enthalpy.					
	$\Rightarrow h_3 = h_4$	(3)				
4-1	An internally, reversible, constant prise vaporated to a saturated vapor a evaporation is supplied by the reframount of heat transferred to the refrigeration load.	ressure heat interaction in which the working fluid at state point 1. The latent enthalpy necessary for rigerated space surrounding the evaporator. The working fluid in the evaporator is called the				

$$\Rightarrow q_1 = h_1 - h_4 \tag{4}$$

The thermal efficiency of the cycle can be calculated as:

$$\eta = \frac{q_{evap}}{w_{comp}} = \frac{h_1 - h_4}{h_{2s} - h_1}$$
(5)

3. Preparation and properties of nanofluid

Multi-walled carbon nanotubes (MWCNT) were prepared from Research Institute of Petroleum Industry (R.I.P.I.) with 90–95% purity were dispersed in distilled water, as the base fluid, to form the CNT–water nanofluids. The average diameter of the nanotubes varies from 10 to 20 nm and their length from 5 to 15 μ n. The nanofluids were synthesized by the two-step method, without any surfactant in order to not affect the viscosity and the thermal conductivity of suspensions. Favorable volume fraction of CNT-water nanofluid was prepared by mixing appropriate quantities of nanoparticles with the water, and then sonicated by an ultrasonic bath (Hielscher UP400S, H40sonotrode) for at least 100 min. The CNT nanofluid used in this study stayed stable for a period of 60h without any visible settlement.



Fig. 3: Typical SEM image of carbon nanotubes.

Bahram Jalili et al. / International Journal of Material Science Innovations (IJMSI) 2(1): 8-17, 2014



Fig. 4: Typical TEM image of carbon nanotubes.

4. Experimental procedure

The refrigeration cycle performance was first measured using pure water as the working fluid for base data. Then, pure water with various mass fractions of CNT particles were used as the working fluid for the same tests. Finally, we compare heat transfer in two different conditions and it was shown the effect of nanoparticles on the refrigeration cycle performance clearly.

5. Experimental apparatus



Fig. 5: Schematic of the experimental setup.

Bahram Jalili et al. / International Journal of Material Science Innovations (IJMSI) 2(1): 8-17, 2014

6. Discussion and results

The refrigeration cycle heat transfer results with pure water were compared to water and CNT particles mixture. According to below figures, It is shown that increasing in CNT particles will be caused great temperature gradient in entry and discharging evaporator respect to pure water. for example in t=592(s), evaporator entry pure water temperature is $10.1(^{\circ}C)$ wheras evaporator entry water/CNT 2000 ppm mixture temperature is $10.8(^{\circ}C)$ and evaporator discharging pure water temperature is $7.6(^{\circ}C)$ wheras evaporator discharging water/CNT 2000 ppm mixture temperature gradient changes from $2.5(^{\circ}C)$ in pure water to $4.3(^{\circ}C)$ in water/CNT 2000 ppm mixture, then temperature increasing results heat transfer increasing. Because of the performance of the cycle has a direct relation with heat transfer in evaporator according to equ.(5), then, mixing of CNT particles with pure water causes increasing in performance of refrigeration cycle. Because of the importance of refrigeration systems in industry, it is suggested using of water/CNT particles mixture as based fluid in refrigeration systems. The main experimental results are shown in tables 2 to 4 and figures 6 to 9. Every test was run 3–5 times at the same conditions to ensure repeatability.

t(s)	$T(^{\circ}C)$	$T(^{\circ}C)$	$T(^{\circ}C)$	$T(^{\circ}C)$	$T(^{\circ}C)$	T (°C)
	Pure water	150 ppm	300 ppm	600 ppm	900 ppm	2000 ppm
0	14.9	14.9	14.9	14.9	14.9	14.9
70	14.1	14.1	14	14	14.1	14
130	13.4	13.4	13.2	13.3	13.4	13.5
190	12.8	12.8	12.6	12.8	12.8	12.7
250	12.3	12.3	12.1	12.3	12.3	12.3
310	11.9	11.8	11.5	11.7	11.7	12
370	11.5	11.4	11.1	11.2	11.3	11.7
430	11	11	10.7	10.9	10.9	11.4
490	10.7	10.7	10.4	10.5	10.6	11.2
592	10.1	10.1	9.8	9.9	10.2	10.8

Table 2: Evaporator entry temperature in pure water and different amount of CNT



Fig. 6: Evaporator entry temperature in pure water and different amount of CNT

t(s)	$T(^{\circ}C)$ Pure water	T(℃) 150 ppm	T (℃) 300 ppm	T (℃) 600 ppm	T (℃) 900 ppm	T (°C) 2000 ppm
0	10.9	11.4	11.1	11.3	11	11
70	10.3	10.8	10.4	10.5	10.4	9.8
130	9.9	10.2	9.8	10.1	9.8	9.1
190	9.4	9.9	9.4	9.6	9.5	8.5
250	9.1	9.4	8.9	9.3	9.2	8.2
310	8.7	9	8.5	8.9	8.7	7.8
370	8.4	8.7	8.2	8.5	8.3	7.4
430	8.1	8.4	7.9	8.1	8.1	7.2
490	7.8	8	7.6	7.8	7.9	7
592	7.6	7.6	7.1	7.4	7.5	6.5

Table 3: Evaporator discharging temperature in pure water and different amount of CNT



Fig.7: Evaporator discharging temperature in pure water and different amount CNT

t(s)	T ($^{\circ}C$) Pure water	т(°С) 150 ррт	т (°С) 300 ррт	T ([°] C) 600 ppm	т(°С) 900 ррт	т (°С) 2000 ppm
0	4	3.5	3.8	3.6	3.9	3.9
70	3.8	3.3	3.6	3.5	3.7	4.2
130	3.5	3.2	3.4	3.2	3.6	4.4
190	3.4	2.9	3.2	3.2	3.3	4.2
250	3.2	2.9	3.2	3	3.1	4.1
310	3.2	2.8	3	2.8	3	4.2
370	3.1	2.7	2.9	2.7	3	4.3
430	2.9	2.6	2.8	2.8	2.8	4.2
490	2.9	2.7	2.8	2.7	2.7	4.2
592	2.5	2.5	2.7	2.5	2.7	4.3

Table 4: Evaporator entry and discharging temperature gradient.



Fig. 8: Evaporator discharging temperature in pure water and different amount of CNT

References

[1] Maxwell JC. Treatise on electricity and magnetism. Oxford: Clarendon Press;1873.

[2] Choi SUS. Enhancing thermal conductivity of fluids with nanoparticles, in Developments and Applications of Non-Newtonian Flows. ASME FED 231/ MD 1995; 66: 99–103.

[3] Dreizin EL. Metal-based reactive nanomaterials. Progress in Energy and Combustion Science 2009; 35(2): 141–67.

[4] Joseph Lik Hang Chau, Chih Chun Kao. Microwave plasma synthesis of TiN and ZrN nanopowders. Materials Letters 2007; 61(7):1583–7.

[5] Hayashi C, Oda M. Research and applications of nano-particles in Japan. Journal of Aero solid Science 1998; 29: 757–60.

[6] Granqvist CG, Buhrman RA. Ultrafine metal particles. Journal of Applied Physics 1976; 47: 2200.

[7] Gleiter H. Nanocrystalline materials, Program. Material Science 1989; 33: 223–315.

[8] Neikov OD. Nanopowders, handbook of non-ferrous metal powders; 2009. p. 80-101.

[9] Fissan HJ, Schoonman J. Vapor-phase synthesis and processing of nanoparticle materials (nano): a European Science Foundation (ESF) program. Journal of Aero solid Science 1998; 29: 755.

[10] Akoh H, Tsukasaki Y, Yatsuya S, Tasaki A. Magnetic properties of ferromagnetic ultrafine particles prepared by a vacuum evaporation on running oil substrate. Journal of Crystal Growth 1978; 4: 495–500.

Bahram Jalili et al. / International Journal of Material Science Innovations (IJMSI) 2(1): 8-17, 2014

[11] Biercuk BJ, Llaguno MC, Radosavljevic M, Hyun JK, Johnson AT. Carbon nanotube composites for thermal management. Applied Physics Letters 2002; 80: 2767–2772.

[12] Ju S, Li ZY. Theory of thermal conductance in carbon nanotube composites. Applied Physics Letters 2006; 353:194–197.

[13] Yao N, Wang ZL, editors. Handbook of microscopy for nanotechnology. Boston: Kluwer Academic Publishers; 2005.

[14] Daniel MC, Astruc D. Gold nanoparticles: assembly, supramolecular chemistry, quantum-size-related properties, and applications toward biology, catalysis, and nanotechnology. Chemical Reviews 2004; 104:293–346.

[15] Trindade T, O'Brien P, Pickett NL. Nanocrystalline semiconductors: synthesis, properties, and perspectives. Chemical Materials 2001; 13: 3843–3858.

[16] Rajamathi M, Seshadri R. Oxide and chalcogenide nanoparticles from hydrothermal/ solvothermal reactions. Current Opinion Solid State Material Science 2002; 6:337–345.

[17] Hulteen JC, Martin CR. In: Fendler JH, editor. Nanoparticles and nanostructured films: preparation, characterization and applications. New York: Wiley; 1998.

[18] Xuan Y, Li Q. Heat transfer enhancement of nanofluids. International Journal of Heat and Fluid Flow 2000; 21: 58–64.

[19] L. Zhang, Q. Ni, Y. Fu, T. Natsuki, One-step preparation of water-soluble single-walled carbon nanotubes, Appl. Surf. Sci. 255 (15) (2009) 7095–7099.